Master of Science in Cognitive Neuroscience & Neuroimaging

Magnetic Resonance Imaging Part 1

Alain Pitiot, Ph.D.



school of psychology university of nottingham

- MRI in 5mn
- spin & magnetic resonance
- B₀ & magnets
- magnet safety



• Unique Selling Proposition

non-ionizing radiation (non-invasive) good spatial 3D resolution superior soft-tissue contrast in vivo extremely versatile (structure, function, blood flow, etc.)

Timeline

Bloch & Purcell (1946)	magnetic resonance phenomenon (interaction atom/radio waves)
Lauterbur (1971)	MRI in a test tube sample (back projection approach)
Mansfield (1973)	k-space approach

EPI (1977), fMRI (1992), DTI (1985)

why & when

• Synopsis of MRI

- 1. subject placed inside intense homogeneous magnetic field: protons align
- 2. transmit radio waves: protons get excited
- 3. turn off radio waves: protons relax
- 4. measure radio waves emitted by protons
- 5. modify field and repeat step 2

Main factors of MRI

quantum properties of nuclear spin radio-frequency excitation, tissue relaxation magnetic field strength and gradient MR sequence (timing of gradient & RF pulses, signal detection)



J. Hornak website

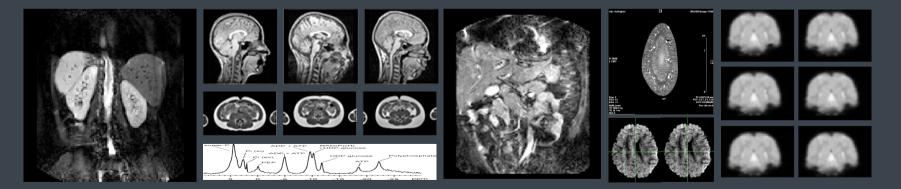
http://www.cis.rit.edu/htbooks/mri/inside.htm

Article

R.A. Pooley, "Fundamental Physics of MR Image", RadioGraphics 2005

Book

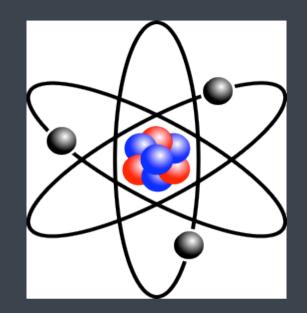
M.A. Bernstein, "Handbook of MRI Pulse Sequences", Academic Press, 2004



MRI in 5mn

building block of matter

 nucleus (protons + neutrons) and cloud of orbiting electrons electrons responsible for chemical reactions



- atom:10⁻¹⁰m, nucleus: 10⁻¹⁴m
 mostly empty
- electrical charges:
 - proton: +, electron: -, neutron: no charge atoms are neutral \leftarrow n^b protons = n^b electrons

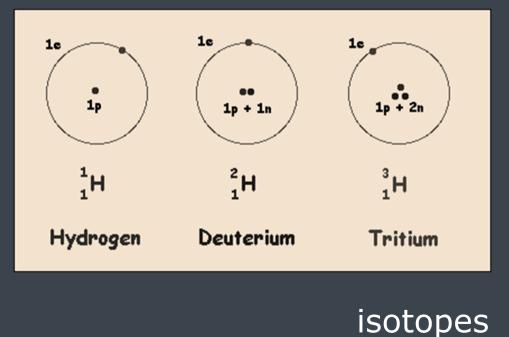
the atom

some notations

atomic number Z: number of protons mass number A: number of nucleons (protons + neutrons)

• isotope = same Z, different A

- almost identical chemical properties
- vastly different
 nuclear properties



- spin ≡ fundamental property of nature multiples of ½, + or – unpaired protons, neutrons, electrons have spin ½ particles with opposite sign spins can pair up: resulting spin is zero (e.g. He)
- nuclei with spin
 - nucleons fill orbitals (like electrons)
 - 2, 8, 20, 28, 50, 82, and 126 nucleons
 - filled orbitals have zero spin

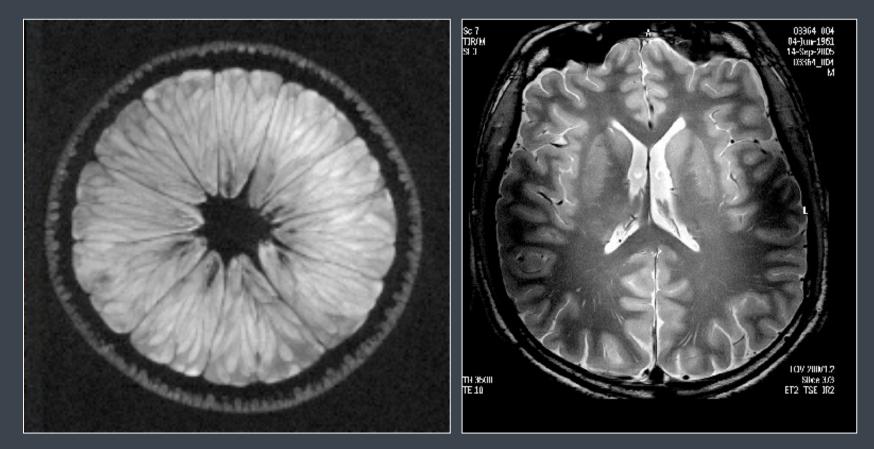
 MR interested in sufficiently abundant isotopes with non zero spin odd number of proton and/or neutron

spin

Nuclei	Unpaired protons	Unpaired neutrons	Net spins	γ (MHz/T)
¹ H	1	0	1/2	42.58
² H	1	1	1	6.54
³¹ P	1	0	1/2	17.25
²³ Na	1	2	3/2	11.27
¹⁴ N	1	1	1	3.08
¹³ C	0	1	1/2	10.71
¹⁹ F	1	0	1/2	40.08

isotopes of interest

¹H, very abundant

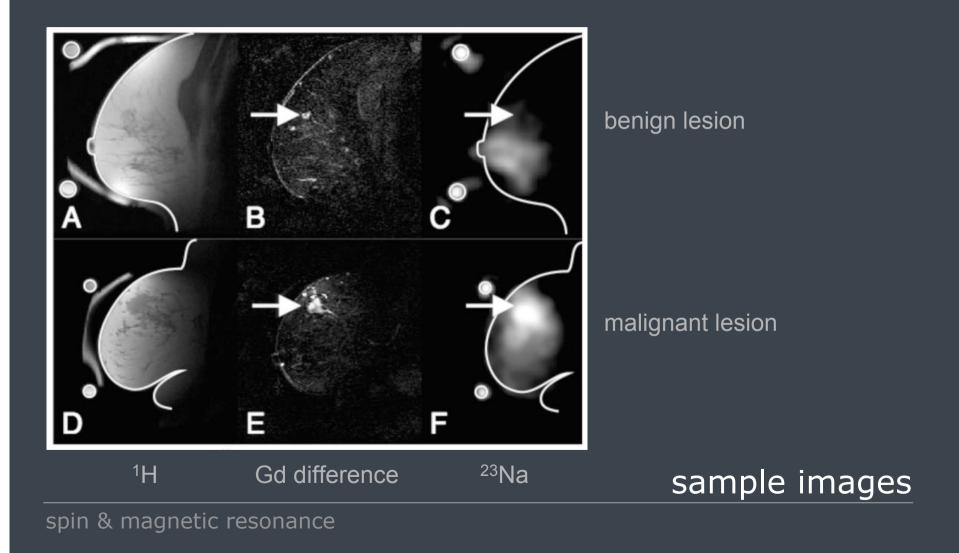


images acquired on SPMMRC 7T scanner

sample images

²³Na, physiology, cellular metabolism

²³Na MRI: From Research to Clinical Use R. Ouwerkerk and R.H. Morgan, J Am Coll Radiol. 2007

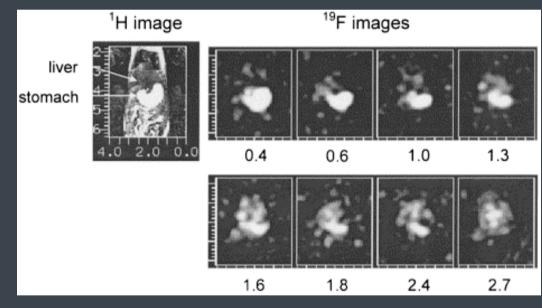


• ¹⁹F, used a tracer

e.g.: perfluorooctanoic acid

¹⁹F Magnetic resonance imaging of perfluorooctanoic acid encapsulated in liposome for biodistribution measurement

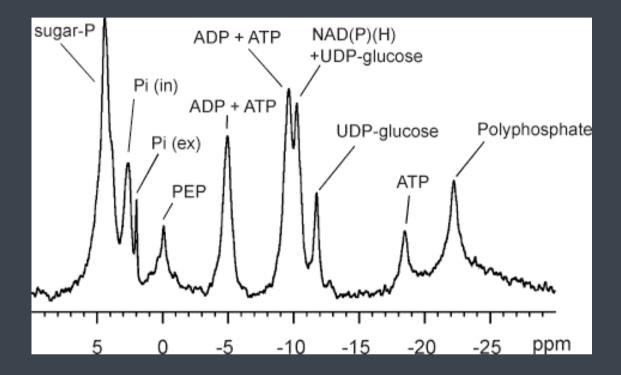
A. Kimura, M. Narazakib, Y. Kanazawab and H. Fujiwaraa Magnetic Resonance Imaging, 2004



in vivo images of PFOA administered mouse

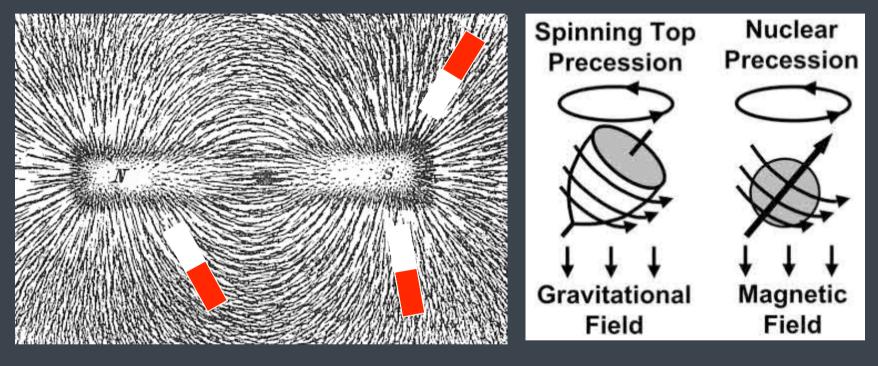
sample images

• ³¹P, energy metabolism





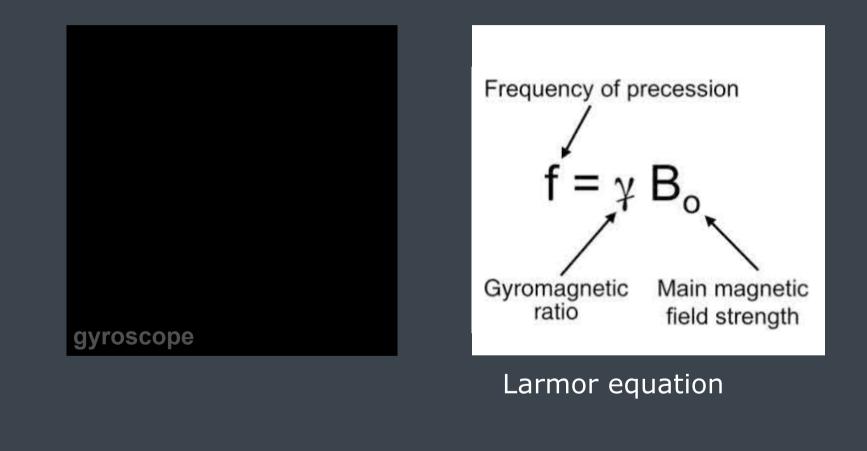
sample images



non spinning magnets align with magnetic field

spinning magnets precess in field

magnetic moment

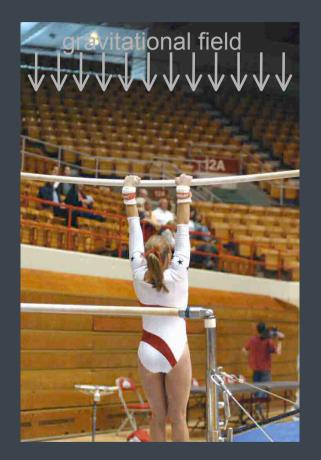


magnetic moment

two energy configurations for gymnast in gravitational field

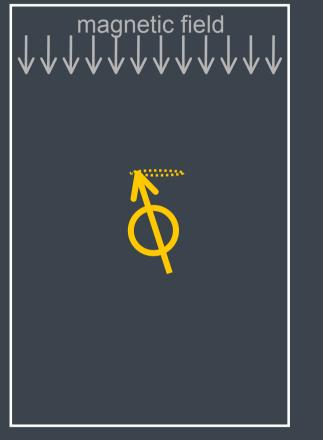


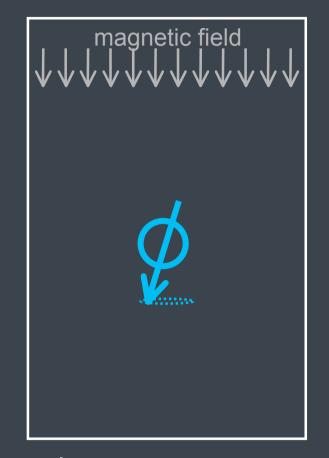
high energy state



low energy state energy level & transition

two energy configurations for proton in magnetic field (Zeeman effect)

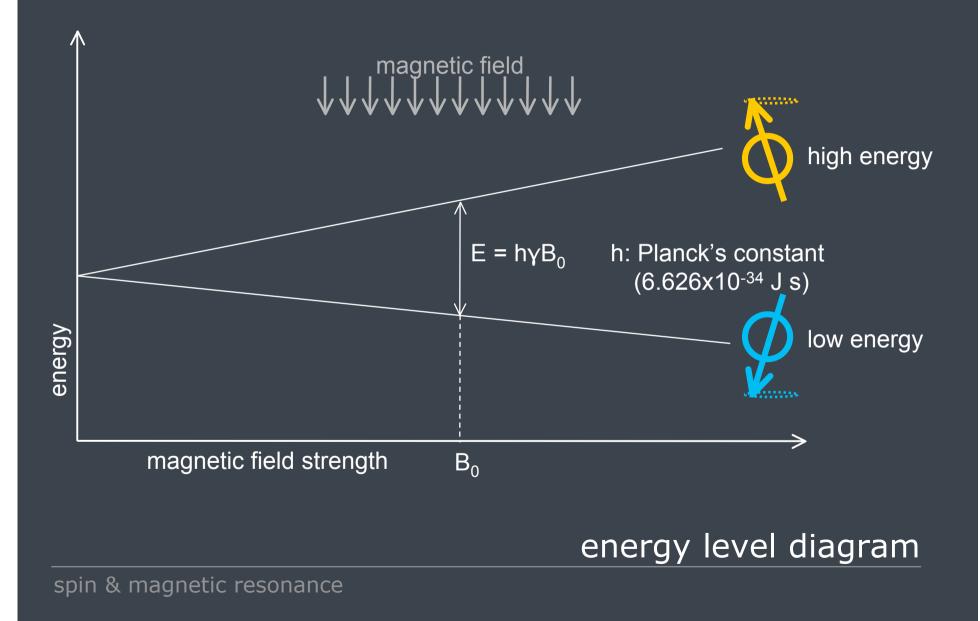




high energy state

low energy state energy level & transition

transition between level by absorption of photon



Zeeman effect is tiny

Boltzmann statistics

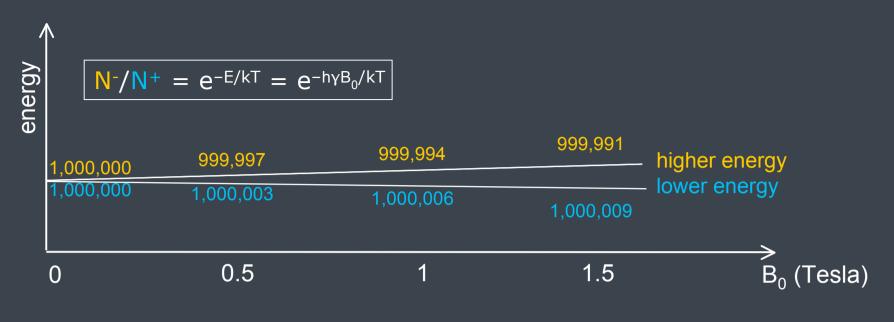
N⁺: number of spins at lower energy level N⁻: number of spins at higher energy level

 $N^{-}/N^{+} = e^{-E/kT} = e^{-h\gamma B_{0}/kT}$

k: Boltzmann constant (1.3805x10-23 J/Kelvin) T: temperature (in Kelvin)

MR signal proportional to population difference between states

Boltzmann statistics



• $B_0 = 1T$, T=310K: 3ppm aligned

• T [: N⁻/N⁺ [

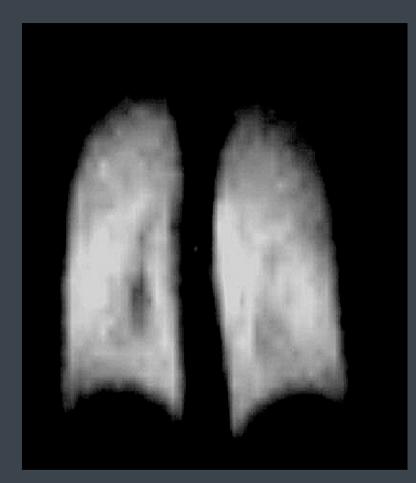
• B₀=0T, T=310K: N⁻/N⁺=1

0% spin aligned no net magnetization



T=0K: N⁻/N⁺=0
 100% spin aligned

field strength, temperature & spin state



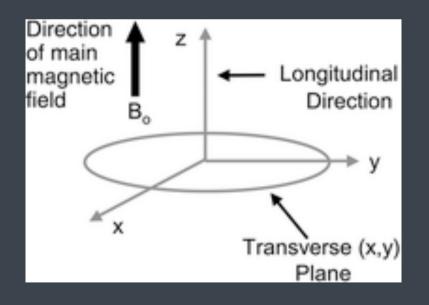
Lung imaging of ³He at 0.15T

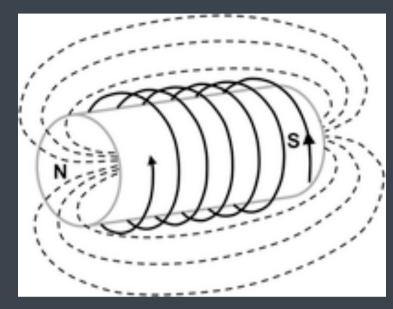
hyperpolarization

$B_0 \equiv$ static field associated with NMR

- created by superconducting magnets
- order of 1T (10⁴ Earth's field)

⇐ coordinate system

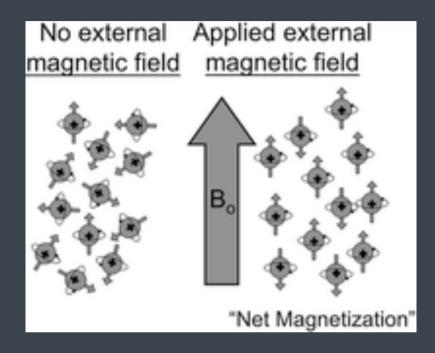






B₀ & magnets

• In B_0 , spins align \leftarrow bulk magnetisation



- spins precess at 54° to the z axis with random phase
- net bulk magnetisation parallel to B₀

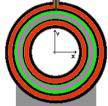


B₀ & magnets

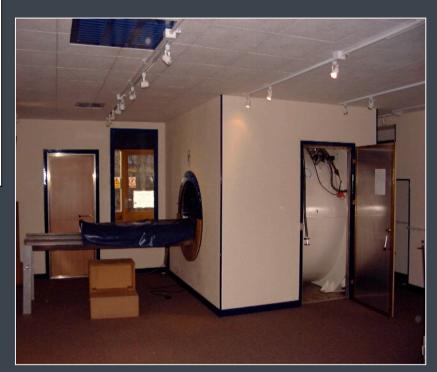
3T Philips Achieva @ SPMMRC

- 60,000 x Earth's field
- superconducting wires
- contains 1000 litres of liquid He





Vacuum
 Liquid Helium
 Liquid Nitrogen
 Container & Support
 Superconducting Coil



superconducting magnets

now

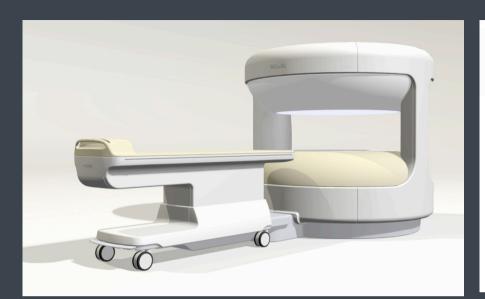
location	1990's	2000′s
district general hospitals	0.5T	1.5T
medical schools	1.5T	3Т
large research institutes	3T	7T

and up to 17T in animals!

(near) future

 11.7T planned at Neurospin (CEA, France) (900mm warm bore)

magnetic strength trends



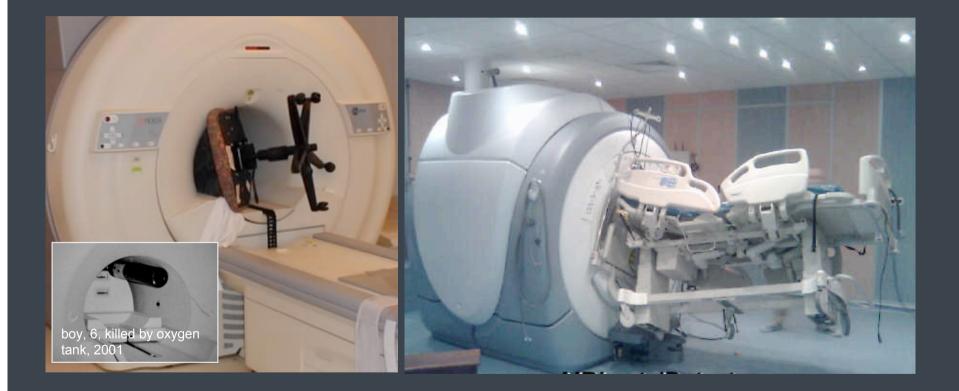




magnet designs

 B_0 & magnets

Magnetic force is powerful, permanent, increases rapidly in proximity of scanner





magnet safety





NO METALLIC IMPLANTS NO NEUROSTIMULATORS

Persons with pacemakers, neurostimulators, or metallic implants must not enter this area. Serious injury may result.





magnet safety



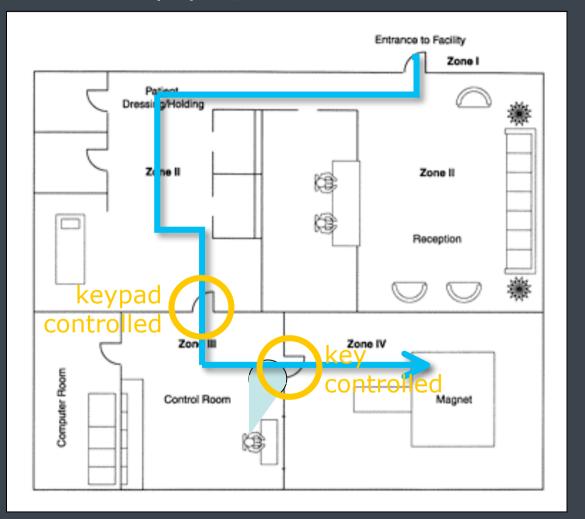
- Classify staff and train appropriately
- Zone spaces to prevent accidental access visually indicate 5 G line for each magnet
- MR incompatible implants and projectiles make no assumptions
 devices that are safe at 1.5T may not be at 3T

screen, screen and screen again

- in writing
- at dressing room, verbally
- screen again on entrance to magnet hall verbally

site management best practice

Limiting access ACR white paper, 2002



site management best practice

magnet safety